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Electrophoretic Au NPs Deposition on Carbon Nanotube Networked Layers for Gas Sensors

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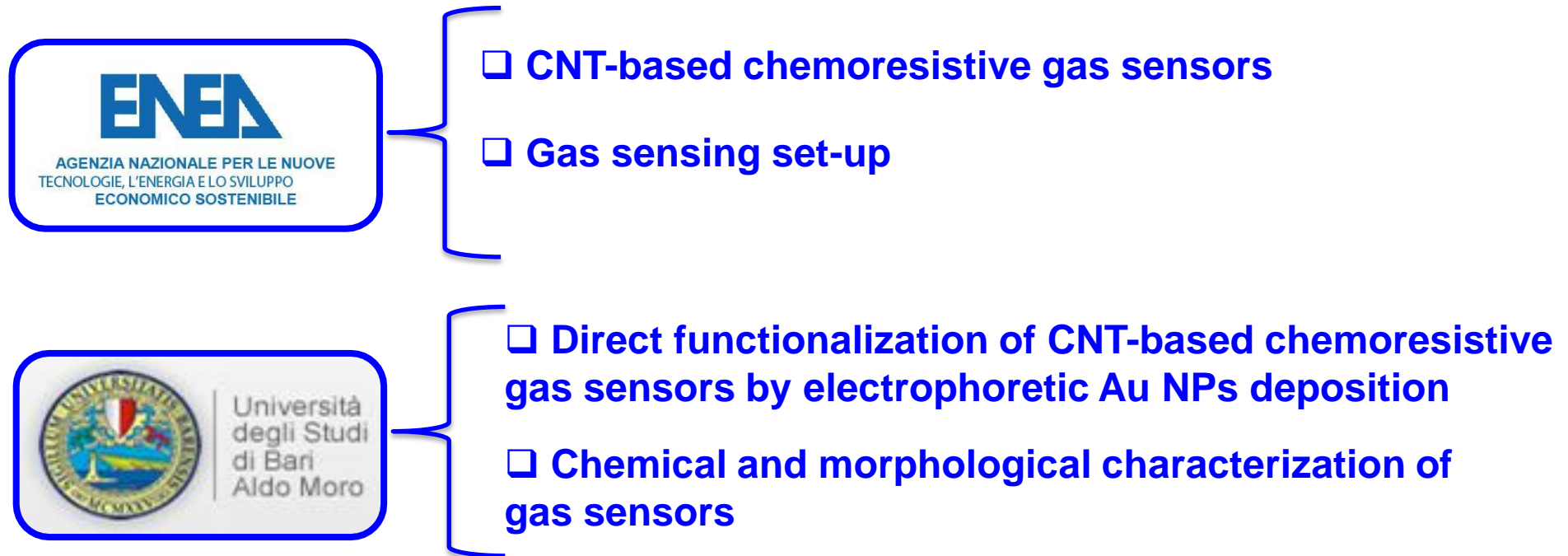
Background - Motivation

- **Involved in 2 National Projects** concerning the synthesis of Automotive Exhaust Gas Sensors based on catalytic metal nanoparticles supported on nanostructured materials for monitoring NO_x, HCs.
- **Background:** Electrochemical Synthesis of colloidal metal nanoparticles; Sol-gel synthesis of nanostructured metal oxides (MO_x); Electrochemical modification of nanostructured materials by metal nanoparticles.

Current research activities

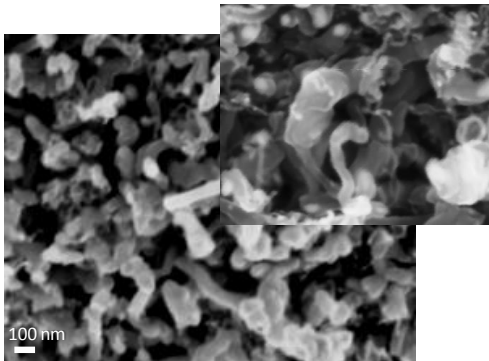
ELECTROCHEMICAL FUNCTIONALIZATION OF CNT-BASED GAS SENSORS BY METAL NPs TO IMPROVE GAS SENSING PROPERTIES

Research collaboration with the sensor laboratory of Dr Michele Penza and the thin film laboratory of Dr Marco Alvisi @ENEA in Brindisi



Experimental set-up

■ CNTs GROWTH By CVD TECHNOLOGY @ ENEA THIN FILM LABORATORY

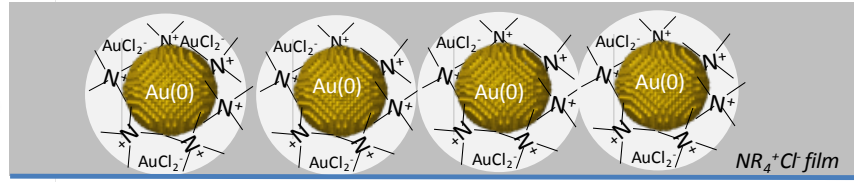
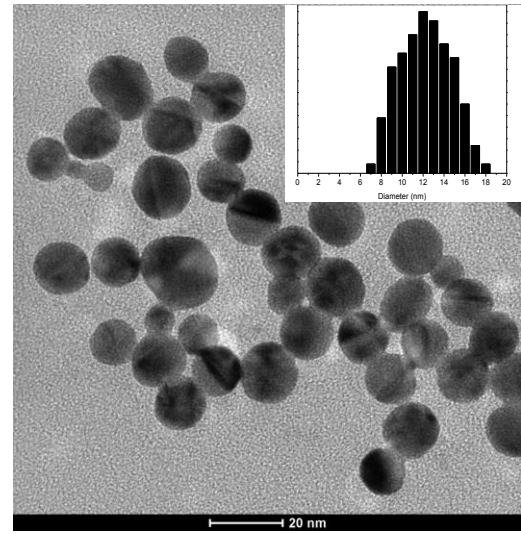
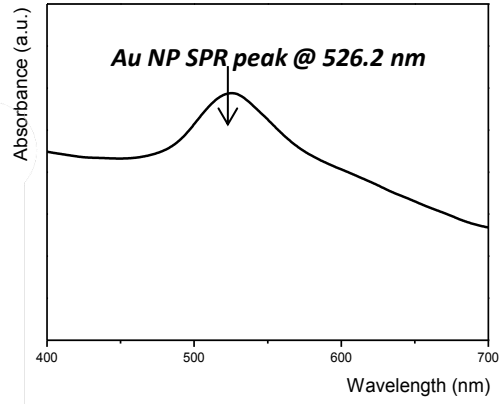
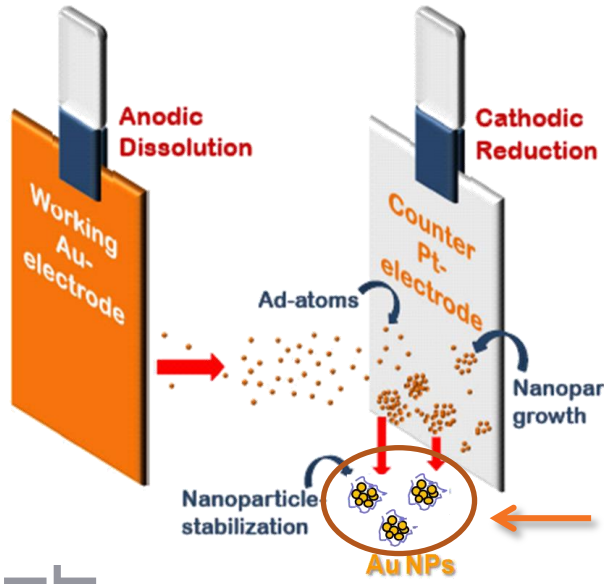
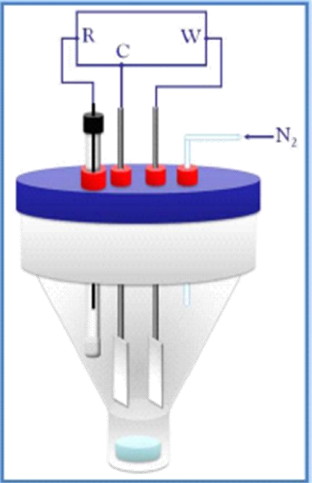


Substrate: film cobalt (Co) nanoclusters, 6 nm thick
1st step: H₂ plasma pretreatment @ 550°C
2nd step: H₂ + C₂H₄ @ 550°C

M. Penza et al., Sens. Actuators B 144 (2010) 387-394

■ Au-NPs electrochemical synthesis Sacrificial Anode Electrolysis (SAE) @ UNIBA

- **Electrolytic solution :**
0.05 M TOAC in THF/ACN=3:1
- **Electrolysis potential:** 1 V
- **Electrolysis charge:** 300 C



Experimental set-up

Electrophoretic deposition of Au NPs on CNTs-based gas sensor device @ UNIBA

➤ Cathodic process $E(V) = -0.5 V$ (ocp: $-0.3 V$)

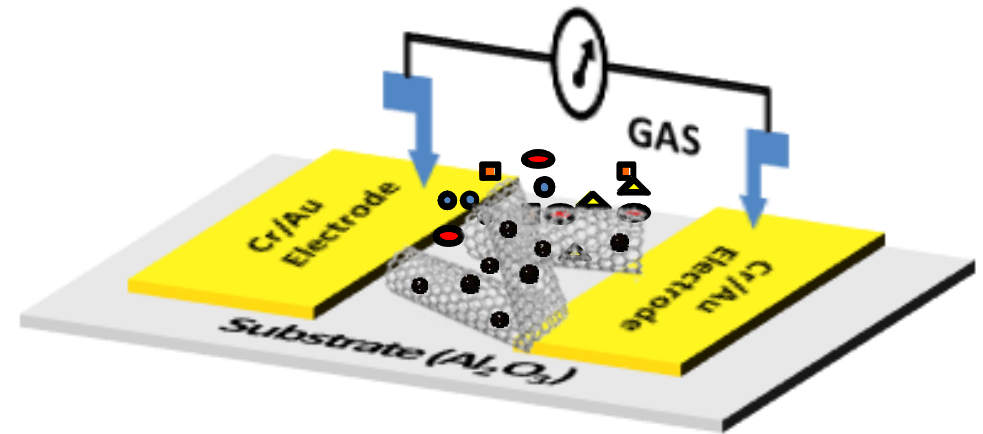
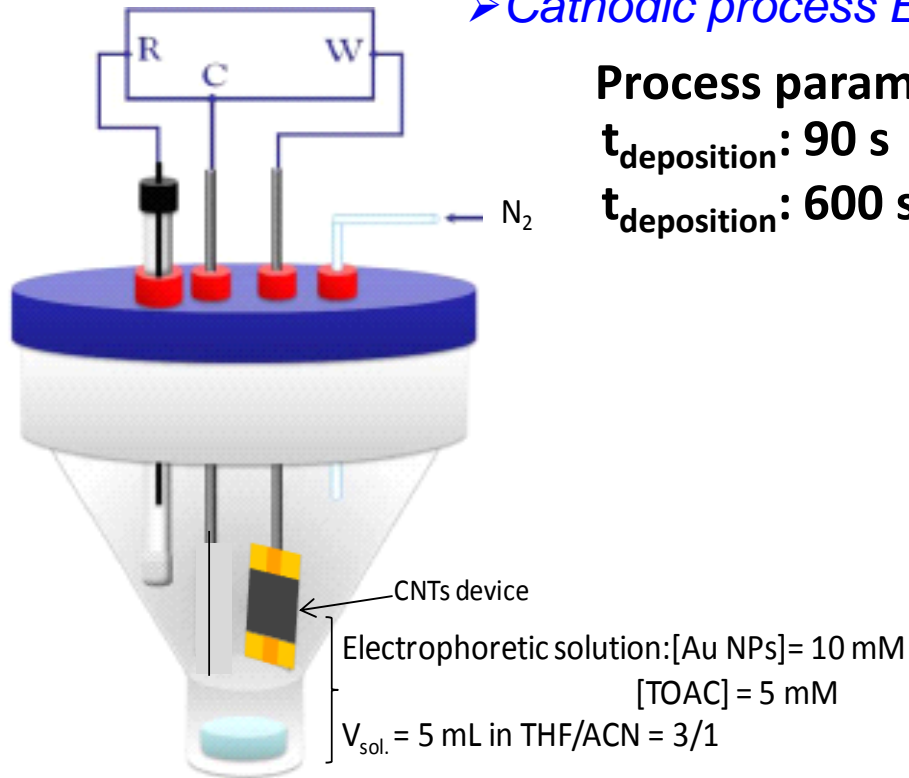
Process parameters:

$t_{\text{deposition}}: 90 \text{ s}$

$t_{\text{deposition}}: 600 \text{ s}$

$i = 6.3 \cdot 10^{-4} \text{ A}$ $Q = 5.598 \cdot 10^{-2} \text{ C}$

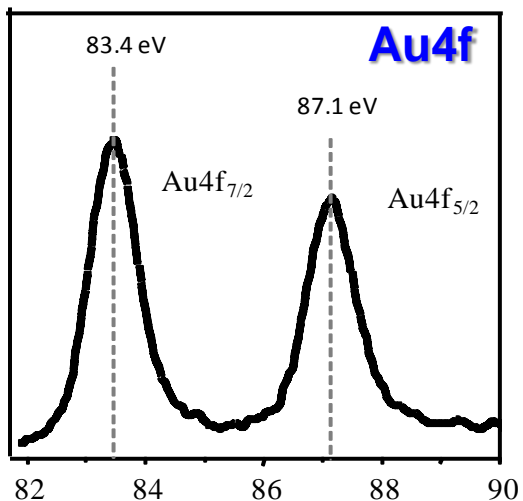
$i = 3.4 \cdot 10^{-4} \text{ A}$ $Q = 2.069 \cdot 10^{-1} \text{ C}$



E. Dilonardo et al., J. Sens. Sens. Syst. 3 1–8 2014.

Characterization

Surface Chemical Analysis by XPS @UNIBA

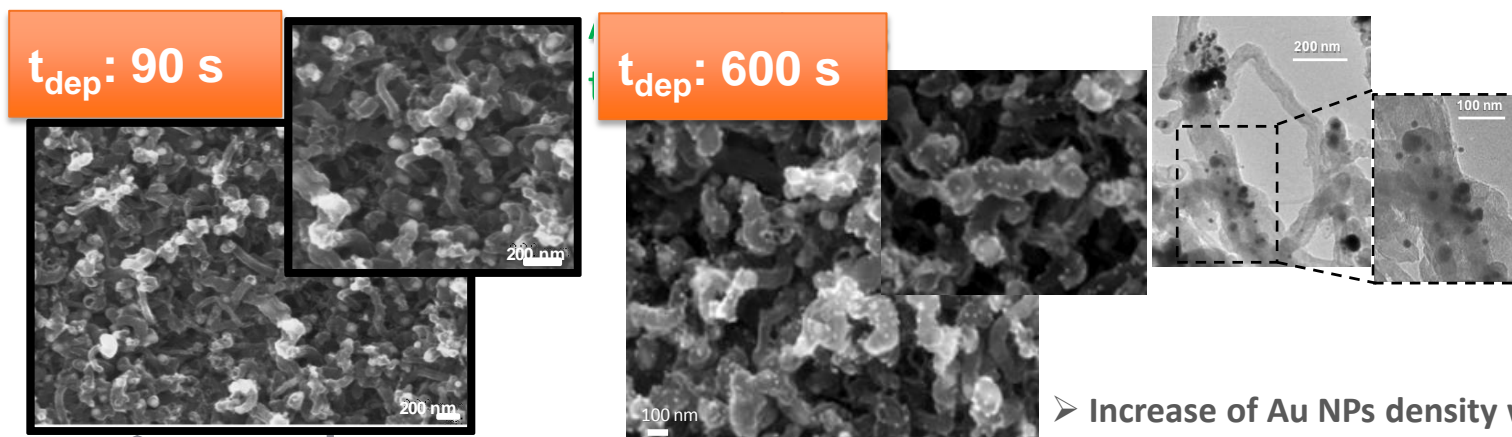


Binding Energy (eV)

SAMPLE	C (%)	Au (%)	O (%)
CNTs as received	95.0 ± 0.5	/	5.0 ± 0.5
Au NPs/CNTs t:90s	94.4 ± 0.5	0.3 ± 0.2	5.3 ± 0.5
Au NPs/CNTs t:600s	91.2 ± 0.5	1.1 ± 0.2	7.8 ± 0.5

- Successful electrophoretic decoration of CNT by Au NPs.
- Nano size of gold particles: Au4f_{7/2}@ 83.4 eV (initial state size effects).
- Fine tuning of Au content by deposition time: increasing of Au loading with the deposition time.

Morphological and Structural analyses @UNIBA



- Increase of Au NPs density with the deposition time.

Characterization

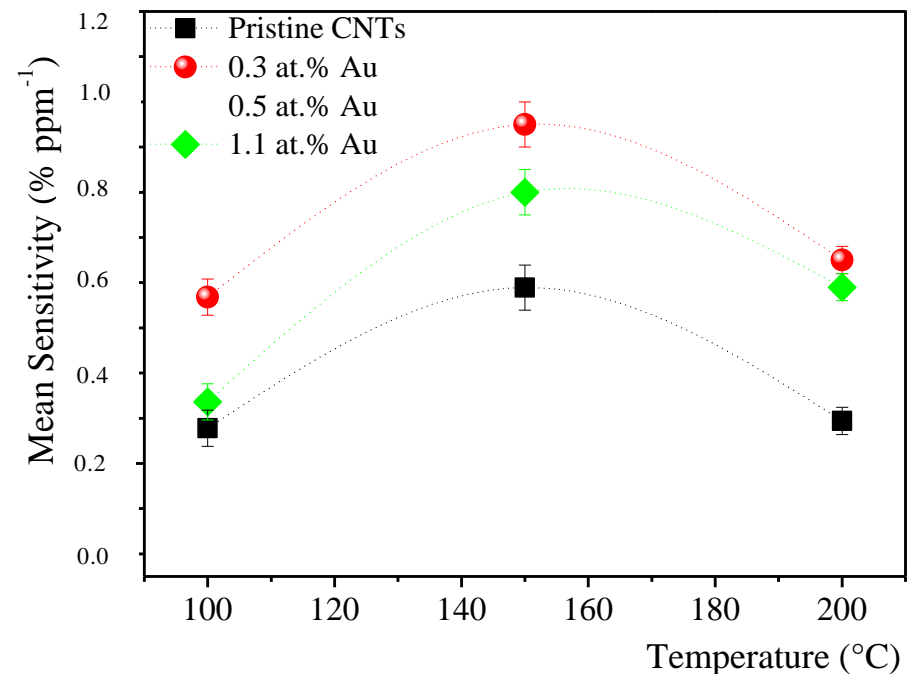
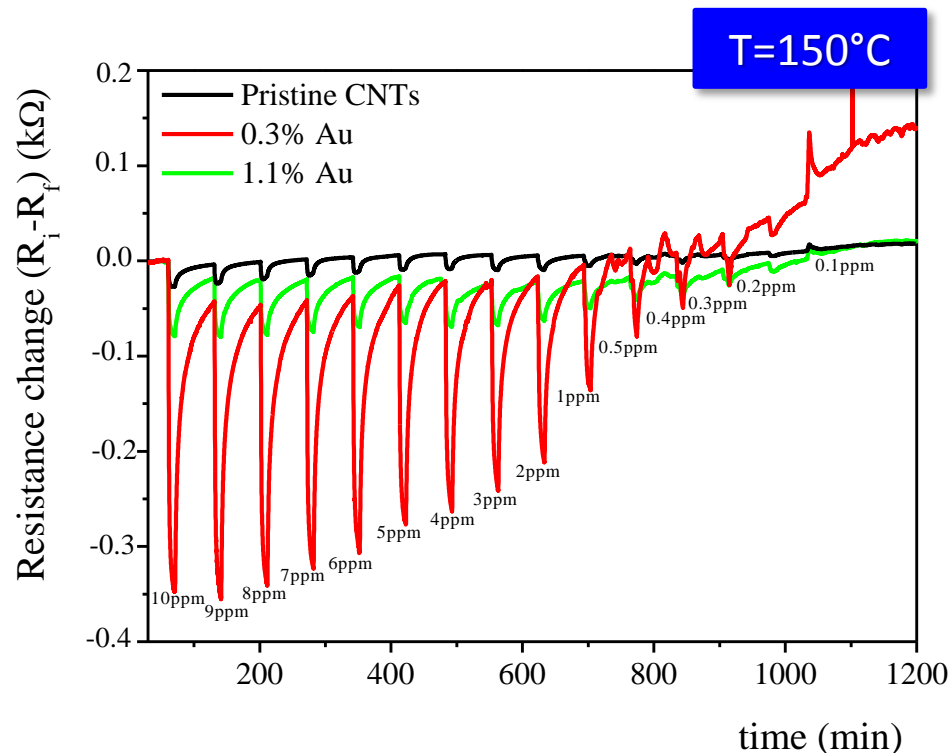
Gas Sensing Measurements @ ENEA

Gas: NO_2 - Carrier Gas: Air

t_{exposure} : 10 min - t_{recovery} : 60 min (in Air)

➤ NO_2 concentration effect [range: 10-0.1 ppm]

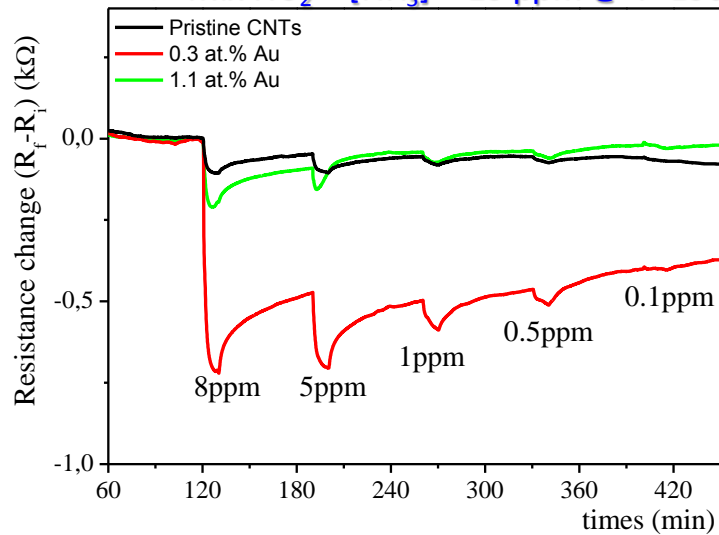
➤ T_{process} effect [range 100-200°C]



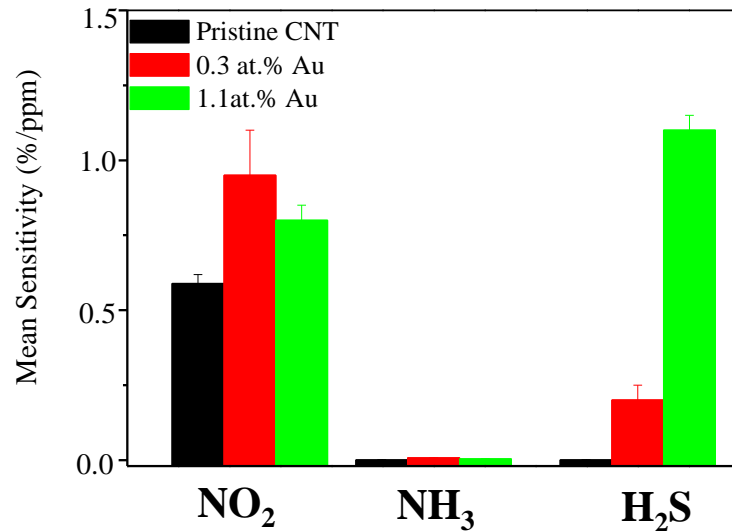
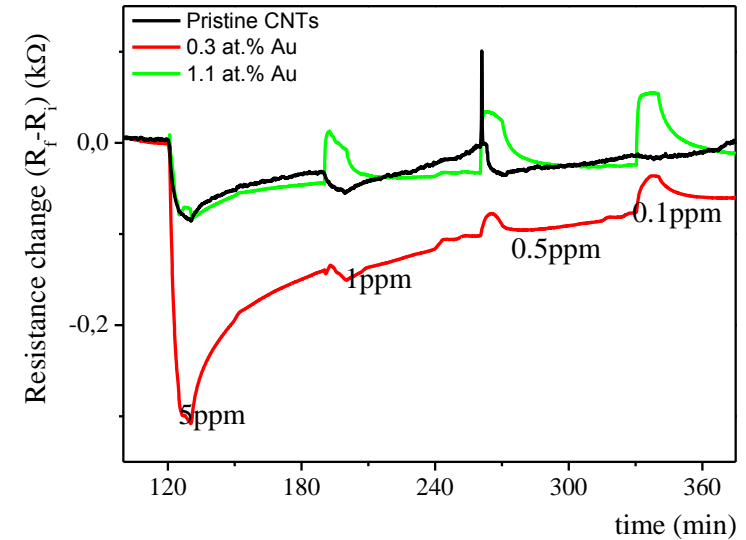
- ✓ NO_2 MEAN SENSITIVITY IS HIGHER FOR Au NPs DECORATED CNTs AT ALL INVESTIGATED T
- ✓ AuNPs DECORATED CNTs ARE STABLE IN THE INVESTIGATED RANGE OF T
- ✓ MAXIMUM NO_2 MEAN SENSITIVITY @ T= 150°C

INTERFERING GASES

➤ Mix $\text{NO}_2 + [\text{NH}_3] = 25 \text{ ppm}$ @ $T=150^\circ\text{C}$



➤ Mix $\text{NO}_2 + [\text{H}_2\text{S}] = 1 \text{ ppm}$ @ $T=150^\circ\text{C}$



- ✓ High selectivity for NO_2 at low Au content
- ✓ High selectivity for H_2S at high Au content



CONCLUSION

- ✓ A tunable loading of Au NPs with uniform dimension is efficiently deposited directly on the surface of CNTs-based sensor device by electrophoretic process.
- ✓ Au NPs functionalized CNTs-based gas sensor have a higher thermal stability than un-functionalized one.
- ✓ Functionalized CNTs-based gas sensor with low Au content reveals a higher NO₂ sensitivity and selectivity, also for [NO₂] in sub-ppm range.

FUTURE PROSPECTIVES

- ✓ Investigation of the gas sensing mechanism of Au-functionalized CNTs towards NO₂ and H₂S.
- ✓ Electrophoretic functionalization of CNTs-based gas sensor devices with other metals and/or metal oxides nanoparticles.

Acknowledgments

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