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## DEVELOPING AIR QUALITY SENSORS BY LASER DEPOSITION ON GRAPHENE



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## **Outline**

- Motivation
- Fabrication of graphene sensors
- Sensibilisation by PLD (pulsed laser deposition)
- Benchmarking NO<sub>2</sub> sensors
- Explorations for other gases
- Conclusions



## **Motivation**

- High potential of <u>graphene</u> fully exposed to environment; responses to single gas molecules have been demonstrated
- For unlocking the potential of graphene new approaches are required for increasing the sensitivity in <u>real atmospheric</u> <u>measurements</u> and for making the devices selective to <u>different</u> <u>target gases</u>
- A fruitful technique is the <u>sensibilisation</u> of single layer graphene by pulsed laser deposition (PLD). Depending on PLD target material and process parameters, adsorption centres with different properties can be created at the defects, impurities, and phase boundaries.



## **CVD growth of graphene**





## **Fabrication of sensors**



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## **Pristine CVD graphene**



In ambient conditions the response is almost absent or <1% level



# Sensibilisation by pulsed laser deposition (PLD)



- Target can be any solid material
- Particle kinetic energies can be varied between 0.025 and ~1000 eV
- Typical deposition rates -1% of a monolayer per laser pulse



## **PLD facility**



## KrF laser

- 248 nm
- 25 ns
- 1 50 Hz
- 2-7 mJ/cm<sup>2</sup>

## Process control by ellipsometry and plasma spectrometry



## **PLD processes**

The process was carried out in

- vacuum (<10<sup>-6</sup> mbar)
- oxygen or nitrogen gas at 10<sup>-2</sup>...5x10<sup>-2</sup> mbar



Number of laser pulses 1...3000

Deposition materials (targets):

- Oxides (NiO,  $ZrO_2$ ,  $SnO_2$ ,  $TiO_2$ ,  $V_2O_3$ )
- Metals (Ag, Au, Pd, Ru)

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## Raman spectra

P=10<sup>-6</sup> mbar

P=10<sup>-2</sup> mbar







## From single defects to porous nanostructures



## **Graphene/ZrO**<sub>2</sub>

## NO<sub>2</sub> in air at RT



## ~50 times higher response after PLD



# Stability of NO<sub>2</sub> sensor on graphene/PLD(TiO<sub>2</sub>)





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**Benchmarking (I)** 

14

## **Benchmarking (II)**



Highly selective: responses <1% for CO (100 ppm) and SO<sub>2</sub> (5 ppm)





Increased sensitivity towards NO<sub>2</sub> was obtained in case of all PLD targets/processes tested. Look at electron affinities:

- NO<sub>2</sub> 2.273 eV
- SO<sub>2</sub> 1.107 eV
- O<sub>2</sub> 0.45 eV

Any hope for other gases?



## Sensing NH<sub>3</sub> with graphene/PLD(SnO<sub>2</sub>)





## **Responses to different gases**

Oxide/Gas	NO <sub>2</sub>	NH <sub>3</sub>	СО	SO <sub>2</sub>
TiO <sub>2</sub>	Х	х	0	0
SnO <sub>2</sub>	х	Х	0	X*
V <sub>2</sub> O <sub>3</sub>	x	х	Х	х

0 – response (almost) absent, x – 'normal' response, X - largest response. X\* - sensor made by 2-stage deposition.

The gas concentrations were typically at 100 ppb level for  $NO_2$  and  $SO_2$  and at 10 ppm level for CO and  $NH_3$ .

## Sensing CO with graphene/PLD(V<sub>2</sub>O<sub>3</sub>)





## Sensing SO<sub>2</sub> with graphene/PLD(SnO<sub>2</sub>)



2-stage PLD:

 6 high energy pulses – defects with average distance ~4 nm

2) 3000 low energy pulses (in background gas) – ~10 nm thick porous coating

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## CONCLUSIONS

### Main achievements:

- Graphene/PLD NO<sub>2</sub> sensors
  - Prototypes fabricated, calibrated, benchmarked
  - Small influence of humidity
  - Stable over 5 months
- Potential for other gases (NH<sub>3</sub>, CO, SO<sub>2</sub>) demonstrated (for specific PLD targets and process parameters)

## open PROBLEMS:

- Understanding the factors behind selectivity!
- New (simpler) fabrication routes?

## **Collaborators and support**

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